



# TOYOHASHI UNIVERSITY of TECHNOLOGY

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## PRESS RELEASE

Source: Toyohashi University of Technology, Japan, Committee for Public Relations

Release Title: Leap toward robust binder-less metal phosphide electrodes for Li-ion batteries

Release Subtitle: Binder-less tin phosphide/carbon composite film electrodes for lithium-ion batteries, fabricated by aerosol deposition process

### Release Summary Text:

Researchers at the Toyohashi University of Technology have successfully fabricated a binder-less tin phosphide ( $\text{Sn}_4\text{P}_3$ )/carbon (C) composite film electrode for lithium-ion batteries via aerosol deposition. The  $\text{Sn}_4\text{P}_3/\text{C}$  particles were directly solidified on a metal substrate via impact consolidation, without applying a binder. Charging and discharging cycling stabilities were improved by both complexed carbon and controlled electrical potential window for lithium extraction. This finding could help realize advanced lithium-ion batteries of higher capacity.

### Complete text of Release:

Lithium-ion (Li-ion) batteries are widely used as a power source in portable electronic devices. They have recently attracted considerable attention because of their potential to be employed in large-scale as a power source for electric vehicles and plugin hybrid electric vehicles and as stationary energy storage systems for renewable energy. To realize advanced Li-ion batteries with higher energy density, anode materials with higher capacity are required. Although a few Li alloys such as Li–Si and Li–Sn, whose theoretical capacity is much higher than that of graphite (theoretical gravimetric capacity = 372 mAh/g), have been extensively studied, they generally result in poor cycling stability due to the large variation in volume during charging and discharging reactions.

Tin phosphide ( $\text{Sn}_4\text{P}_3$ ) (theoretical gravimetric capacity = 1255 mAh/g) with layered structure, generally used as a high-capacity alloy-based anode material for Li-ion batteries, has an averaged operation potential of  $\sim 0.5$  V vs.  $\text{Li}/\text{Li}^+$ . Reports indicate that complexing carbon materials with nano-structured  $\text{Sn}_4\text{P}_3$  particles significantly enhance the cycling stability. Generally, electrodes used in batteries are fabricated by coating a slurry comprising electrode active materials, conductive carbon additives, and binders on metallic foils. For carbon complexed  $\text{Sn}_4\text{P}_3$  ( $\text{Sn}_4\text{P}_3/\text{C}$ ) anodes (such as those reported in the literature), the weight fraction of the active materials in an electrode is decreased by approximately 60-70 % because of the use of significant quantities of conductive additives and binders to achieve stable cycling. Consequently, the gravimetric specific capacity per electrode weight (including those of conductive carbon additives and binders) is decreased significantly.



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Researchers at the Department of Electrical and Electronic Information Engineering, Toyohashi University of Technology, have successfully fabricated a binder-less  $\text{Sn}_4\text{P}_3/\text{C}$  composite film electrode for Li-ion battery anodes via aerosol deposition (AD). In this process, the  $\text{Sn}_4\text{P}_3$  particles are complexed with acetylene black using simple ball-milling method; the obtained  $\text{Sn}_4\text{P}_3/\text{C}$  particles are then directly solidified on a metal substrate via impact consolidation without adding any other conductive additives or binders. This method enables enhancement of the fraction of  $\text{Sn}_4\text{P}_3$  in the composite to above 80%. Furthermore, structural change of the composite electrode is reduced and cycling stability is improved for both complexed carbon and controlled electrical potential window for lithium extraction reaction. The  $\text{Sn}_4\text{P}_3/\text{C}$  composite film fabricated by the AD process maintains gravimetric capacities of approximately  $730 \text{ mAh g}^{-1}$ ,  $500 \text{ mAh g}^{-1}$ , and  $400 \text{ mAh g}^{-1}$  at 100<sup>th</sup>, 200<sup>th</sup>, and 400<sup>th</sup> cycles, respectively.

The first author Toki Moritaka is quoted as saying, “Although optimizing the deposition conditions was difficult, useful information on enhancement of cycling stability of the  $\text{Sn}_4\text{P}_3/\text{C}$  composite film electrode fabricated by the AD process was obtained. The complexed carbon functions not only as a buffer to suppress the collapse of electrodes caused by the large variation in volume of  $\text{Sn}_4\text{P}_3$ , but also as an electronic conduction path among the atomized active material particles in the composite.”

“This process is an effective means to increase the capacity value per electrode weight. We believe there is scope for improvement of the electrochemical performance by the size and content of the carbon in  $\text{Sn}_4\text{P}_3/\text{C}$  used in composite film fabrication by the AD process. We are now trying to optimize the complexed carbon content and increase the composite film thickness,” quotes Associate Professor Ryoji Inada.

The findings of this study may contribute to the realization of advanced Li-ion batteries of higher capacity. Moreover, because not only Li but Na can also be stored in and extracted from  $\text{Sn}_4\text{P}_3$  by similar alloying and dealloying reactions, the  $\text{Sn}_4\text{P}_3$  electrode can be employed in next-generation Na-ion batteries at much lower costs.

Funding agency:

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## Reference:

Toki Moritaka, Yuh Yamashita, Tomohiro Tojo, Ryoji Inada, Yoji Sakurai (2019). Characterization of  $\text{Sn}_4\text{P}_3$ -Carbon Composite Films for Lithium-Ion Battery Anode Fabricated by Aerosol Deposition: *Nanomaterials*, 9(7), 1032.10.3390/nano9071032

## Further information

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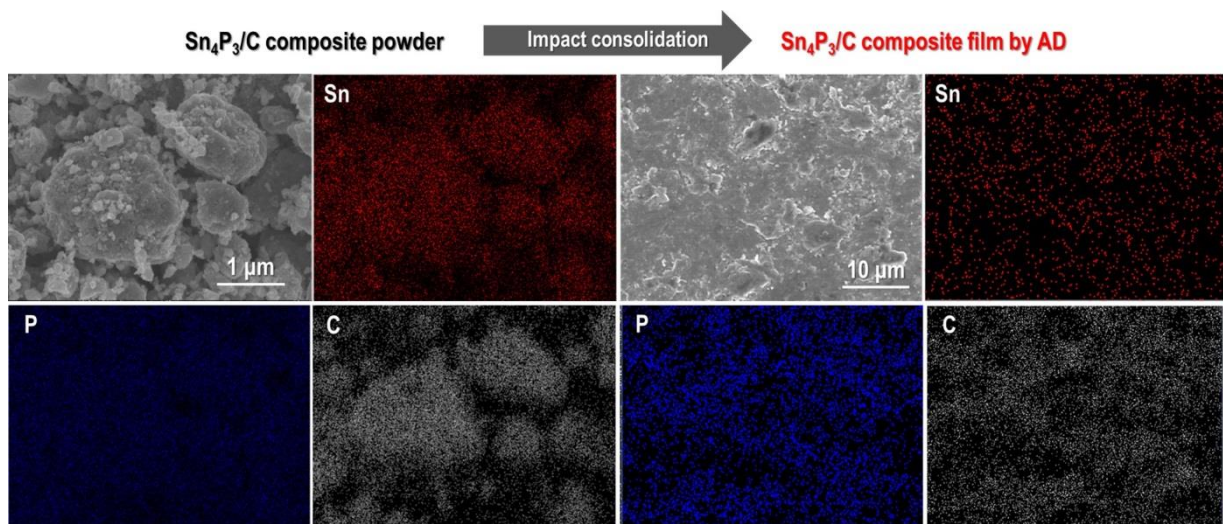
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Toyohashi University of Technology founded in 1976 as a National University of Japan is a research institute in the fields of mechanical engineering, advanced electronics, information sciences, life sciences, and architecture.

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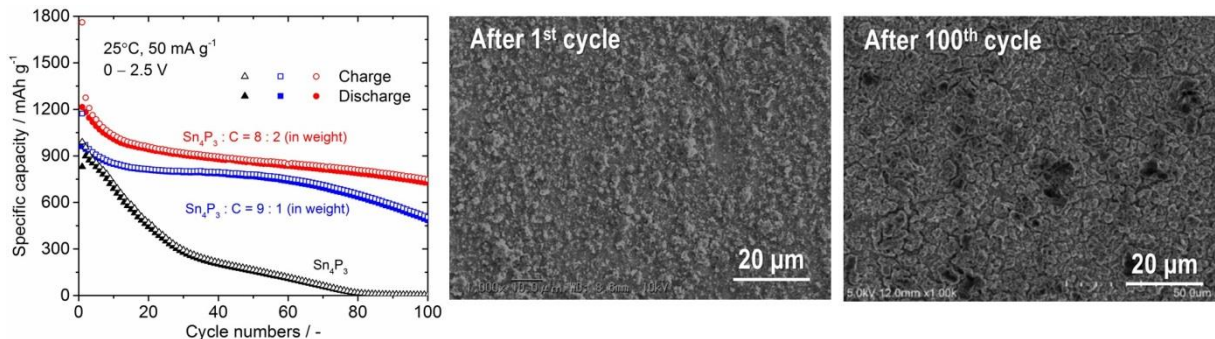
Figure 1:



Caption: Scanning electron microscope (SEM) images of  $\text{Sn}_4\text{P}_3/\text{C}$  composite particles (1<sup>st</sup> row:1<sup>st</sup> image), and surface of  $\text{Sn}_4\text{P}_3/\text{C}$  composite film fabricated by the AD process (1<sup>st</sup> row:3<sup>rd</sup> image). Corresponding elementary distributions for Sn, P, and C are also shown.

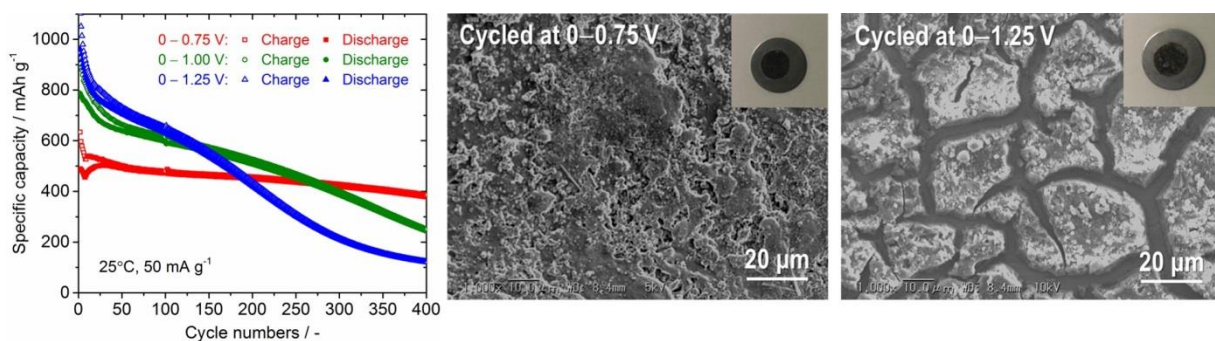


Figure 2:



Caption: Cycling performance of  $\text{Sn}_4\text{P}_3$  and  $\text{Sn}_4\text{P}_3/\text{C}$  composite films for different carbon contents (left) in a two-electrode cell with Li metal as the counter electrode. Cycling test was performed at cell voltage windows ranging from 0 V to 2.5 V. SEM images of  $\text{Sn}_4\text{P}_3/\text{C}$  composite film surface after the 1<sup>st</sup> (middle) and 100<sup>th</sup> cycles (right) are also shown.

Figure 3:



Caption: Long-term cycling performance of  $\text{Sn}_4\text{P}_3/\text{C}$  composite films at different cell voltage windows ranging from 0 V to 0.75 V, and that from 1 V and to 1.25 V (left). SEM images of  $\text{Sn}_4\text{P}_3/\text{C}$  composite film surfaces cycled at 0 V–0.75 V (middle) and that at 0 V–1.25 V (right) are also shown.

Primary keyword: CHEMISTRY/PHYSICS/MATERIALS SCIENCES, ELECTRICAL ENGINEERING/ELECTRONICS, ENERGY/FUEL (NON-PETROLEUM), TECHNOLOGY/ENGINEERING/COMPUTER SCIENCE